## POZNAN UNIVERSITY OF TECHNOLOGY



#### EUROPEAN CREDIT TRANSFER AND ACCUMULATION SYSTEM (ECTS)

pl. M. Skłodowskiej-Curie 5, 60-965 Poznań

### **COURSE DESCRIPTION CARD - SYLLABUS**

Course name

Corrosion-resistant alloys

**Course** 

Field of study Year/Semester

Materials Engineering 3/6

Area of study (specialization) Profile of study

general academic

Level of study Course offered in

First-cycle studies polish

Form of study Requirements

full-time elective

**Number of hours** 

Lecture Laboratory classes Other (e.g. online)

15 15

Tutorials Projects/seminars

**Number of credit points** 

2

**Lecturers** 

Responsible for the course/lecturer:

Responsible for the course/lecturer:

Dr inż. Natalia Makuch-Dziarska

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Faculty of Materials Engineering and Technical

**Physics** 

Piotrowo St 3, 60-965 Poznań

**Prerequisites** 

Knowledge: Knowledge of engineering materials, basic chemistry.

Skills: Logical thinking, independent learning, using the library and the Internet.

Social competences: The student is aware of the importance and understands the non-technical aspects and effects of engineering activities, including its impact on the environment.

## **Course objective**

Students receive a complete set of basic information to help understand corrosion phenomena, and are ready to find solutions to corrosive problems with metal materials.

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### **Course-related learning outcomes**

### Knowledge

- 1. The student should have detailed knowledge of K\_W10 engineering materials
- 2. The student should have knowledge of chemistry to understand the structure of elements and chemical compounds; describe the elements of inorganic chemistry, acids, bases, salts, types of reactions, chemical equilibrium, chemical kinetics, electrochemistry, basics of metallurgy [K W03]

### Skills

- 1. The student is able to obtain information from literature, databases and other properly selected sources of materials engineering [K U01]
- 2. The student is able to identify and formulate simple engineering tasks of a practical nature, characteristic of materials engineering, including in particular the selection of materials for specific applications; define the working conditions of the materials. [K U16]
- 3. The student is able to select engineering materials, methods of shaping the structure and properties of materials for technical applications depending on the structure, properties and conditions of use [K U21]

### Social competences

- 1. The student understands the need for lifelong learning; can inspire and organize the learning process of other people [K K01]
- 2. The student correctly identifies and resolves dilemmas related to the profession [K K05]

#### Methods for verifying learning outcomes and assessment criteria

Learning outcomes presented above are verified as follows:

Lecture: written credit at the end of the semester (credit if at least 51% of points are obtained: <51% 2 - ndst, 51% -62% 3 - dst, 63% -74% 3.5 - dst +, 75% -84% 4 - db, 85% -94% 4.5 - db +,> 94% 5 - very good).

Laboratory: Credit based on a written test and written studies on the program content implemented during exercises. In order to pass the written test and all studies must be passed with a positive mark.

#### **Programme content**

#### Lecture:

- 1. Types of corrosion and mechanisms of corrosive destruction of metals and alloys.
- 2. Classification of corrosion-resistant steels. Intergranular corrosion of austenitic corrosion-resistant steels.
- 3. Passivity, passivating metals and non-ferrous alloys.
- 4. Non-ferrous alloys resistant to corrosion.
- 5. Corrosion protection methods.

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6. Corrosion resistance testing methods.

### Laboratory:

- 1. Identification of the corrosion mechanism
- 2. Surface layers and corrosion-resistant coatings
- 3. Corrosion-resistant steels
- 4. Corrosion-resistant nickel alloys
- 5. Corrosion resistance of aluminum alloys. Anodizing of aluminum.

### **Teaching methods**

- 1. Lecture: multimedia presentation
- 2. Laboratory exercises: practical exercises, discussion and teamwork.

### **Bibliography**

#### Basic

- 1. J. Baszkiewicz, M. Kamiński, "Korozja materiałów", Oficyna wydawnicza PW, Warszawa 2006
- 2. S. Mrowec, T. Werber, Korozja gazowa metali, Wydawnictwo Śląsk, Katowice 1975
- 3. M. Orman, A. Golian, Korozja aluminium i jego stopów, Wydawnictwo Śląsk, Katowice 1963
- 4. pod red. K. Darowicki, Procesy korozyjne, Politechnika Gdańska, Gdańsk 2008

## Additional

- 1. L. A. Dobrzański "Podstawy nauki o materiałach i metaloznawstwo" WNT 2002
- 2. R.K. Tredhewey Corrosion, Longman, 1988

# Breakdown of average student's workload

	Hours	ECTS
Total workload	50	2,0
Classes requiring direct contact with the teacher	35	1,0
Student's own work (literature studies, preparation for	15	1,0
laboratory classes/tutorials, preparation for tests/exam, project		
preparation) <sup>1</sup>		

<sup>&</sup>lt;sup>1</sup> delete or add other activities as appropriate